

Total No. of Questions : 6]

SEAT No. :

P1445

[Total No. of Pages : 2

**TE/Insem/APR-135**  
**T.E. Chemical**  
**TRANSPORT PHENOMENA**  
**(2015 Pattern) (Semester - II)**

*Time : 1 Hour ]*

*[Max. Marks : 30*

*Instructions to the candidates:*

- 1) Neat diagrams must be drawn wherever necessary.*
- 2) Figures to the right indicate full marks.*
- 3) Use of logarithmic tables slide rule, mollier charts, electronic pocket calculator and steam tables is allowed.*
- 4) Assume suitable data if necessary.*

**Q1)** Derive the expressions for Momentum flux, velocity profile. maximum velocity. average velocity for a fluid through annulus. **[10]**

OR

- Q2) a)** List down the assumptions for Hagen Poiseuille's Law. **[5]**
- b)** A horizontal annulus is 27 ft long. The outside radius of inner cylinder is 0.495 inch. The inside radius of outside cylinder is 1.1 inch. The fluid density is 80.3 lb/ft<sup>3</sup> and viscosity is 136.8 lb.m/ft.sec. What is the volume rate of flow, the impressed pressure drop is 5.39 psi? **[5]**

**Q3)** Derive the expressions of heat flux and temperature distribution for viscous heat source. **[10]**

OR

**Q4)** Derive the expressions of heat flux and temperature distribution through variable thermal conductivity. **[10]**

**Q5)** Derive the expressions for molar flux. concentration profile and average concentration for diffusion through heterogenous chemical reaction. **[10]**

OR

**P.T.O.**

- Q6)** a) The diffusivity of the gas-pair  $O_2 - CCl_4$  is measured by observing the steady-state rate of evaporation of  $CCl_4$  into a tube containing  $O_2$ . The distance between the  $CCl_4$  liquid level and the top of the tube is 17.1 cm. The total pressure of the system is 755 mm Hg and the temperature is  $0^\circ C$ . The vapor pressure of  $CCl_4$  at that temperature is 33 mm Hg. The cross-sectional area of the diffusion tube is  $0.82 \text{ cm}^2$ . It is found that  $0.0208 \text{ cm}^3$  of  $CCl_4$  evaporates in 10 hour period after steady-state has been attained. What is the diffusivity of the gas-pair  $O_2 - CCl_4$ ? You may assume that the concentration of  $CCl_4$  at the top of the tube is zero and the density of liquid  $CCl_4$  is  $1.59 \text{ g cm}^{-3}$ . **[5]**
- b) List down the steps involved in approach for solving mass transfer problems. **[5]**

**M M M**