Total No of Questions: [6]

SEAT NO. :

[Total No. of Pages :2]

In-Sem Examination T.E. 2015 (Production) Numerical Techniques and Optimization Methods (Semester - II)

Time: 1 Hour

Max. Marks : 30

- Instructions to the candidates: 1) Neat diagrams must be drawn wherever necessary.
 - 2) Figures to the right side indicate full marks.
 - 3) Use of Calculator is allowed.

4) Assume Suitable data if necessary

- Q1) a) Explain with suitable example (i) Truncation error (ii) Conversion error[4]
 - b) The concentration of pollutant bacteria C in the lake decreases with time according to: [6] $C = 75e^{-1.5t} + 20e^{-0.075t}$

Use Bisection method to determine the time required to reduce the bacteria concentration to

15. Range of initial guess is 3 to 6.(Perform four iterations)

OR

Q2) a) The following system of equation was generated by applying mesh current law to an [8] electrical circuit. Using Gauss Seidel method, determine I_1, I_2, I_3

 $60I_1 - 40I_2 = 200$

 $-40I_1 + 150I_2 - 100I_3 = 0$

 $-100I_2 + 130I_3 = 230$

- **b)** What is condition number? What is its significance?
- Q3) The stress τ and the shear strain rate $\dot{\gamma}$ for a pseudo-plastic fluid can be expressed as: [10] $\tau = \mu \dot{\gamma}^n$. Where, μ is the viscosity and n is exponent of shear strain rate. Using following data for 0.5% hydroxethylcellulose in water solution, determine, μ and n.

$\dot{\gamma}(1/s)$	50	70	90	110	130
$\tau (N/m^2)$	6.01	7.48	8.59	9.19	10.21
OR					

Q4) Data for specific volume 'v' and entropy 's' for superheated H_2O at 200 MPa is given in [10] Table below:

$v (m^3/Kg)$	0.103	0.111	0.125	
s (KJ/Kg K)	0.414	0.545	0.766	

Determine entropy at specific volume of 0.108 m³/kg using Lagrange interpolation method.

[2]

t (sec)	0	1	4	6	8
$c (mg/m^3)$	12	22	32	45	58

For an outflow of 0.3 m³/s, estimate the mass of chemical in grams that exit the reactor from 0 to 8 sec. The mass is given as: $M = Q \int_{t1}^{t2} c. dt$. Use Simpson's 1/3 rule.

OR

Q6)	a)	Explain trapezoidal rule for numerical integration.	[5]
	b)	Explain Runge-Kutta methods for solving ordinary differential equation.	[5]

[10]